



Title: *Staphylococcus carnosus*. Study as an alternativo biocollector for metal minerals

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Editorial label ECORFAN: 607-8695

BCIERMMI Control Number: 2022-01

BCIERMMI Classification (2022): 261022-0001

Pages: 13

RNA: 03-2010-032610115700-14

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Introduction



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Mining Legislation

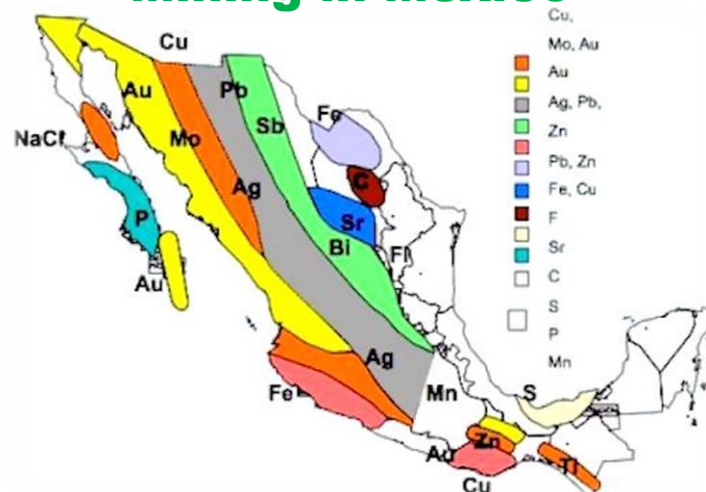


Environmental Mining Economy

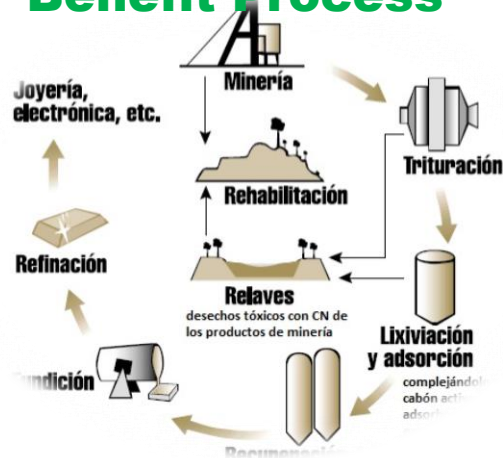


Extraction

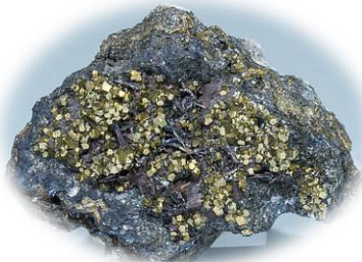
Mining in Mexico



Benefit Process



Chalcopyrite+ Pyrite



Galene + Pyrite

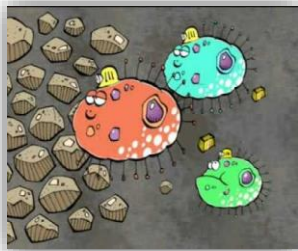
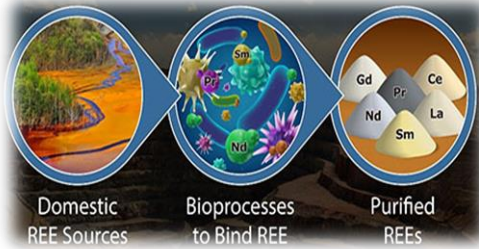
Introduction

Traditional Process



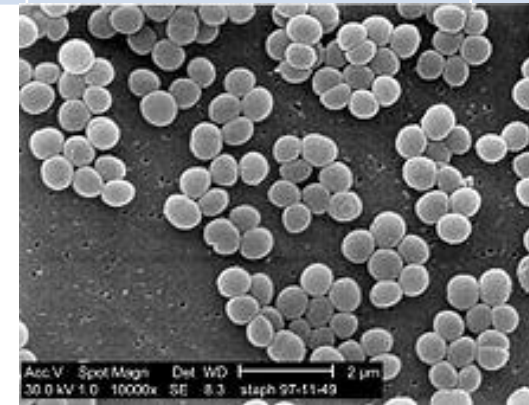
conventional reagents most used in mining
Xanthate

Biomining

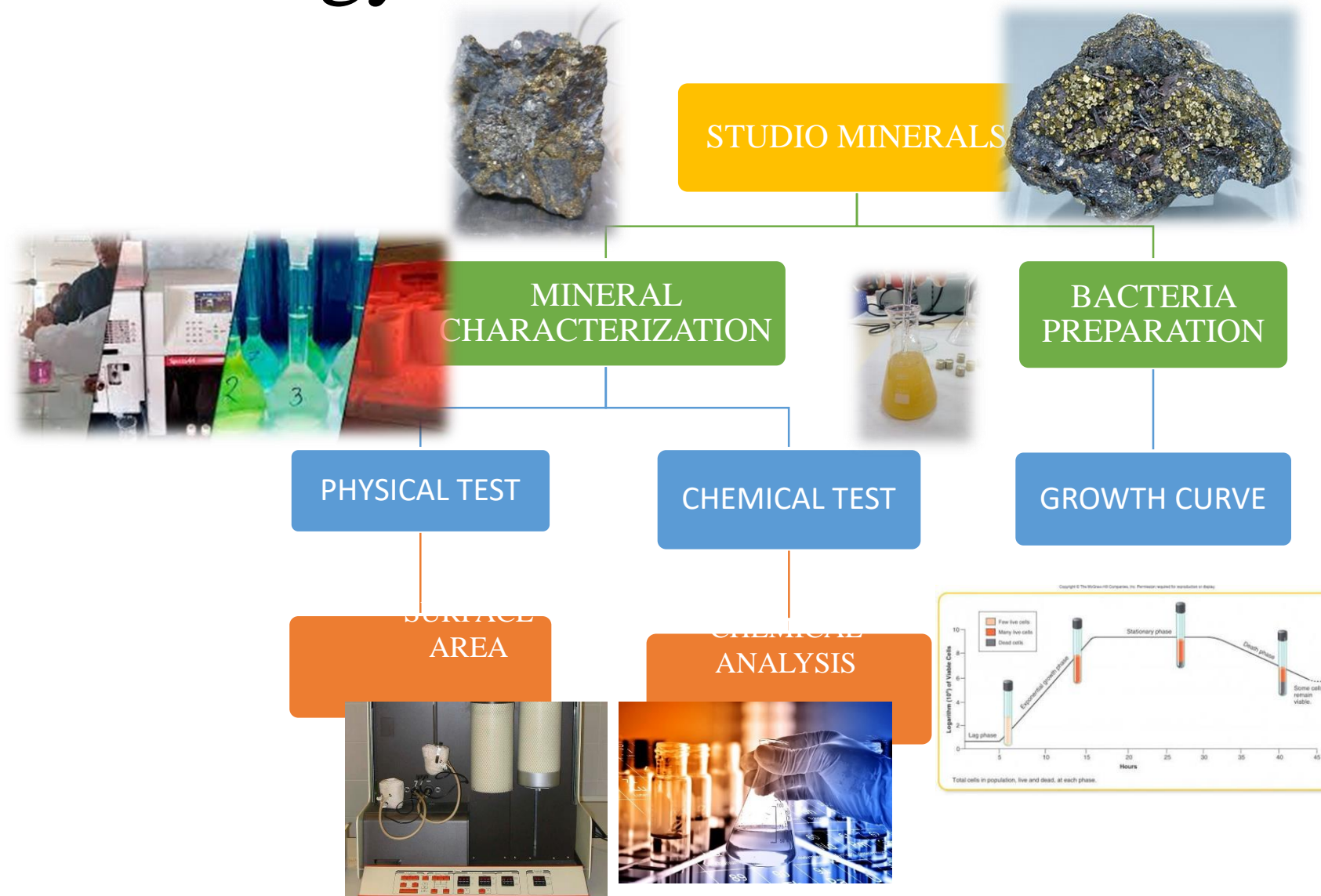


Author	Year	Bacteria	Mineral Studied
Ramos –Escobedo et al	2002	Acidithiobacillus ferrooxidans	Sphalerite and chalcopyrite
Nagaoka <i>et al.</i> , 1999,;	1999	Leptospirillum ferrooxidans,	
Subrahmalannian <i>et al.</i> , 2003,	2003	Polymyxa Paenibacillus,	
Pecina <i>et al.</i> , 2009	2009	Acidithiobacillus thiooxidans	
Santhiya <i>et al</i>	2001 2002	Acidithiobacillus thiooxidans, Acidithiobacillus ferrooxidans	Sphalerite (ZnS)-galena (PbS). Sphalerite . pyrite and arsenopyrite
Ramos-Escobedo et al	2016	Staphylococcus carnosus	Coal

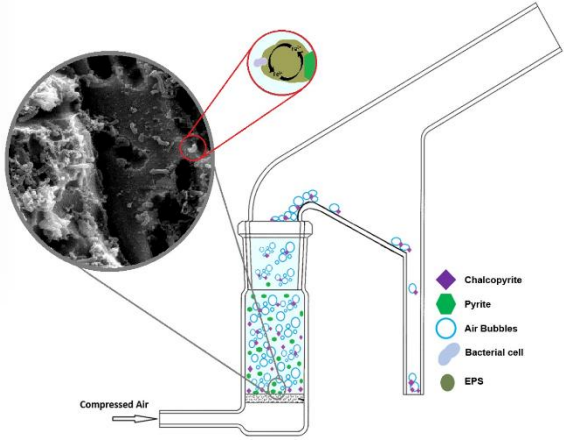
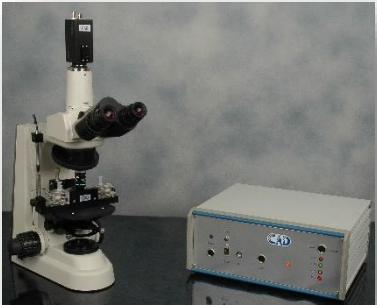
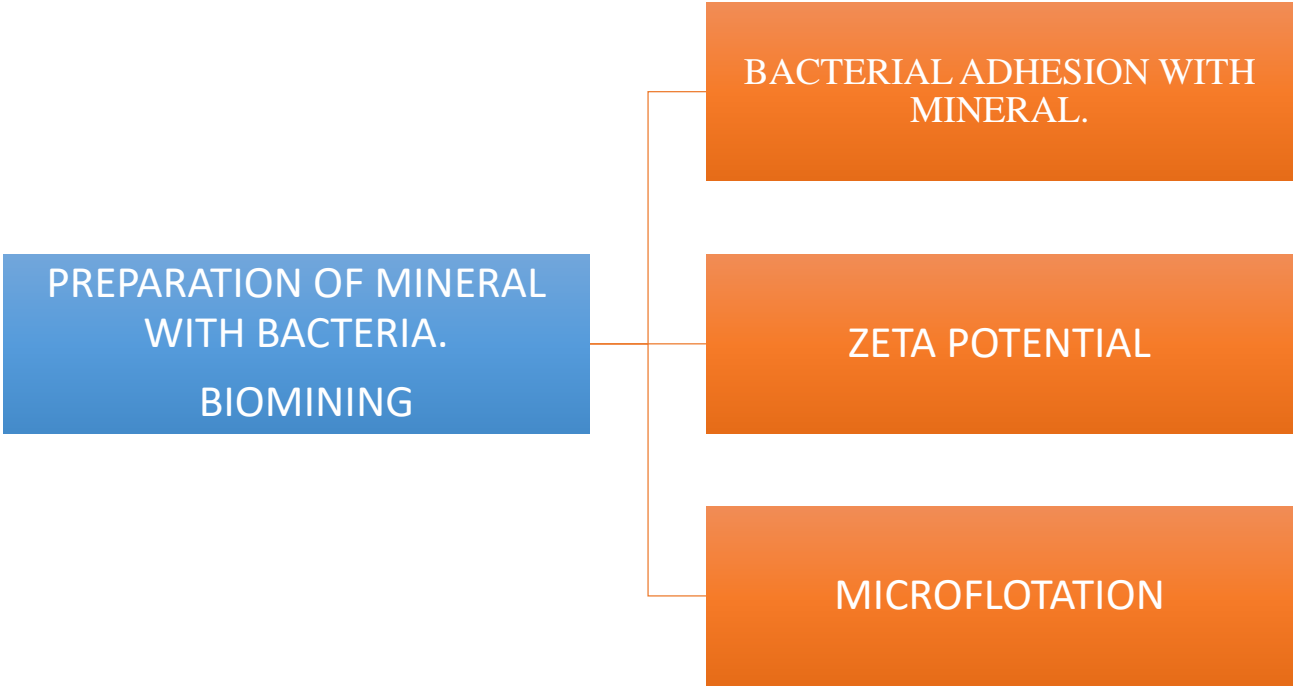
Staphylococcus carnosus



Methodology



Methodology



Results



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Table 1. Chemical analysis of minerals

Minerals	Fórmula	Cu%	Zn%	Fe%	S%	Pb%	Purity %
Chalcopyrite	CuFeS_2	31.24	0.22	32.58	30.45	0.01	90
Galena ⁽⁺⁾	PbS	0.009	0.157	0.135	12.39	73.01	85
Pyrite ⁽⁺⁾	FeS_2	0.002	0.004	40.42	47.439	0.006	88

(+) Associated gangue silicates.

Table 2. Surface area of the mineral fraction used.

<i>Minerals</i>	<i>Surface area, m²/g</i>
<i>Chalcopyrite</i>	<i>0.0109</i>
<i>Pyrite</i>	<i>0.00874</i>
<i>Galena</i>	<i>0.00673</i>

Results

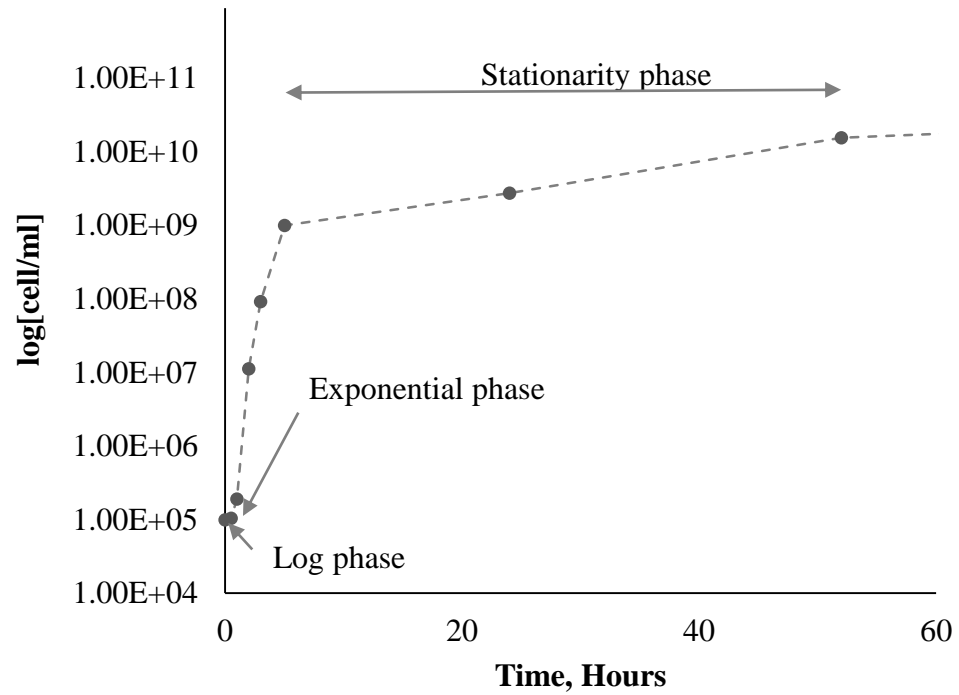


Figure 1. Growth curve of *S. carnosus* bacteria

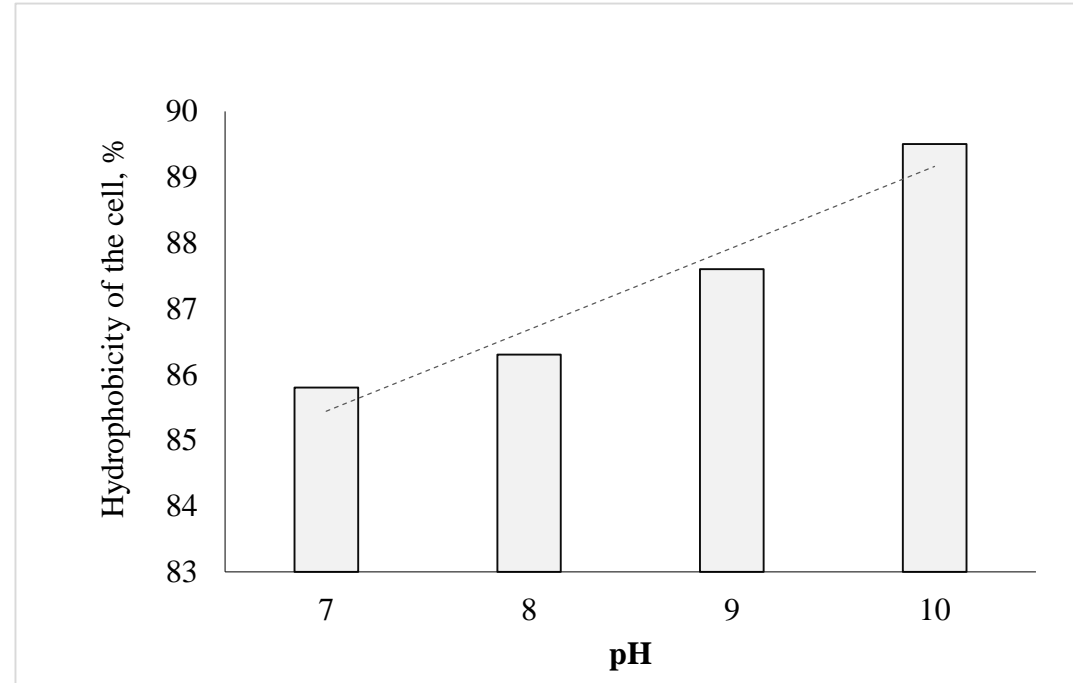


Figure 2. Hydrophobicity of the bacteria as a function of pH

Results



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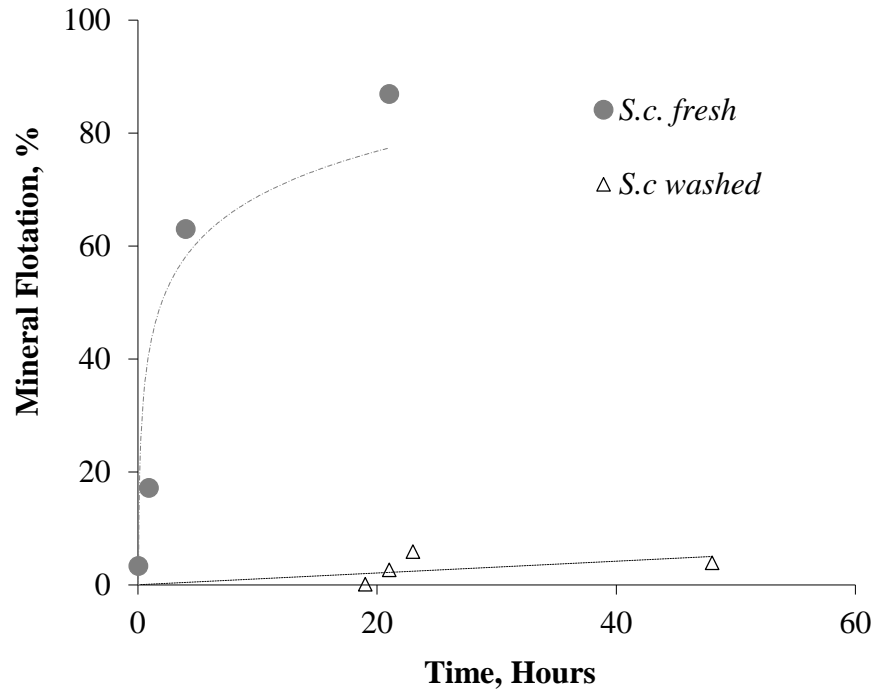


Figure 3. Effect of the culture period the bacteria in the floatability of the chalcopyrite with fresh bacteria (culture medium) and washed bacteria (without culture media).

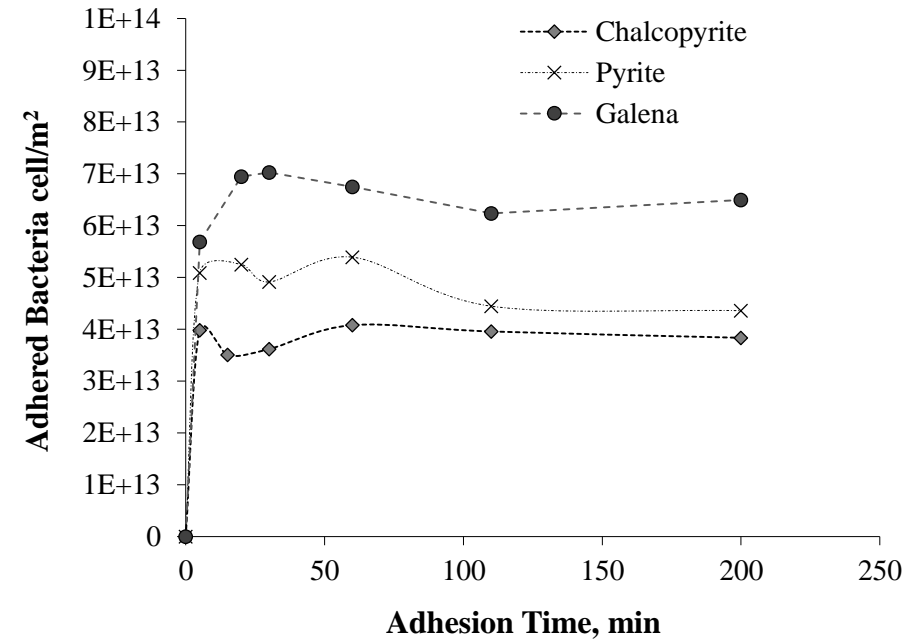


Figure 4. Effect of the floatability of *S. carnosus* + CuFeS_2 bacteria with culture medium and without culture medium at pH 9

Results

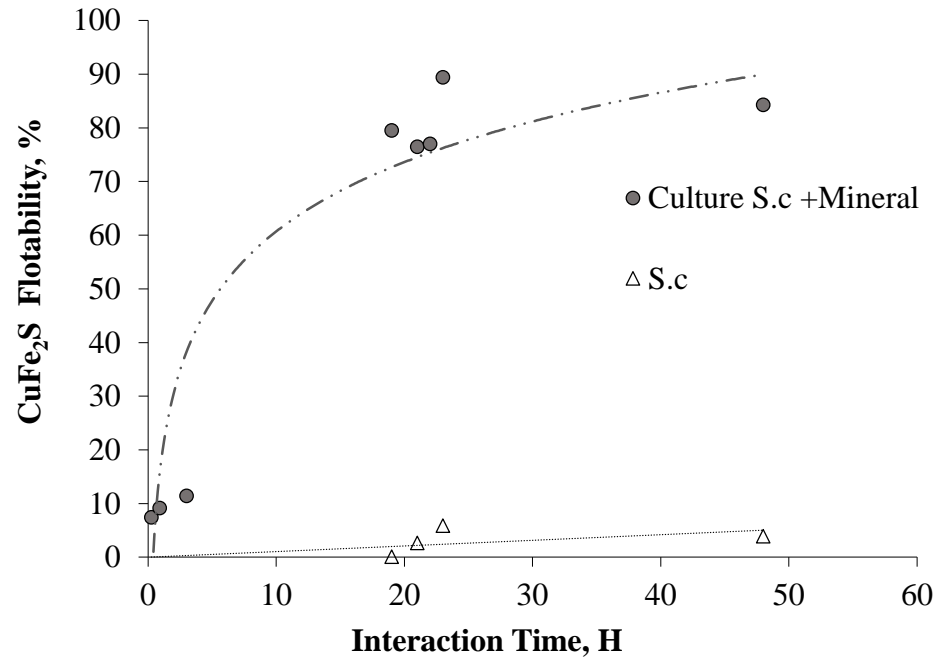


Figure 5. Adhesion isotherms of *S. carnosus* bacteria with minerals of interest at pH 9

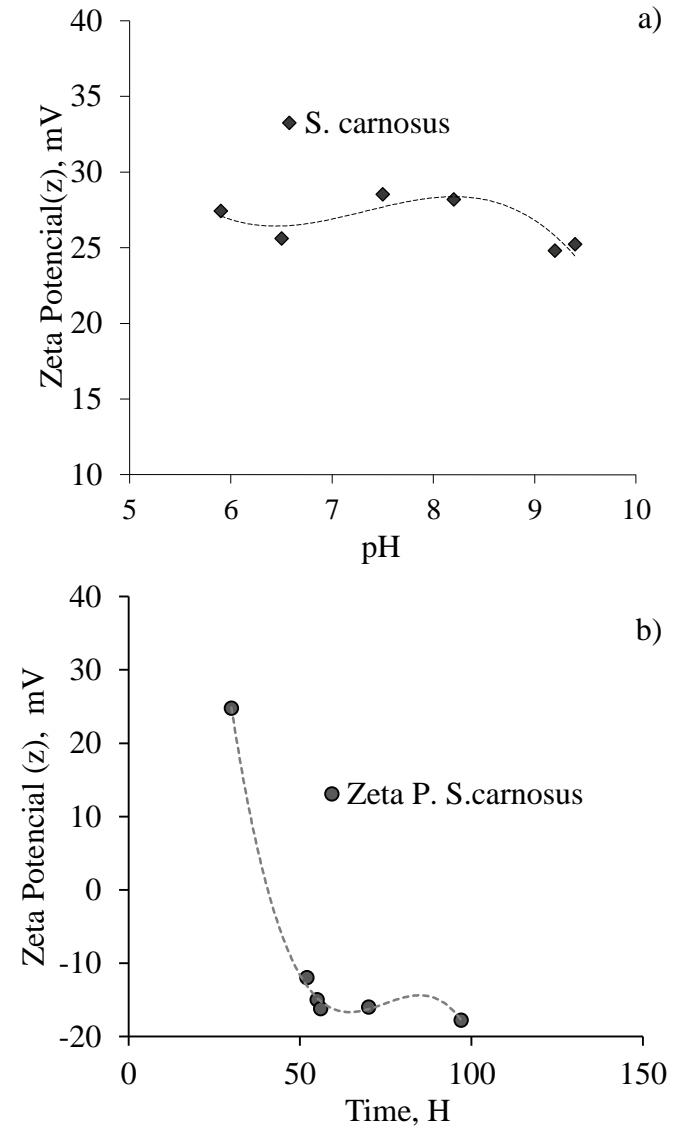


Figure 6. Zeta potential of *S. carnosus* bacteria (control) depending on: a) pH and b) Time (H)

Results



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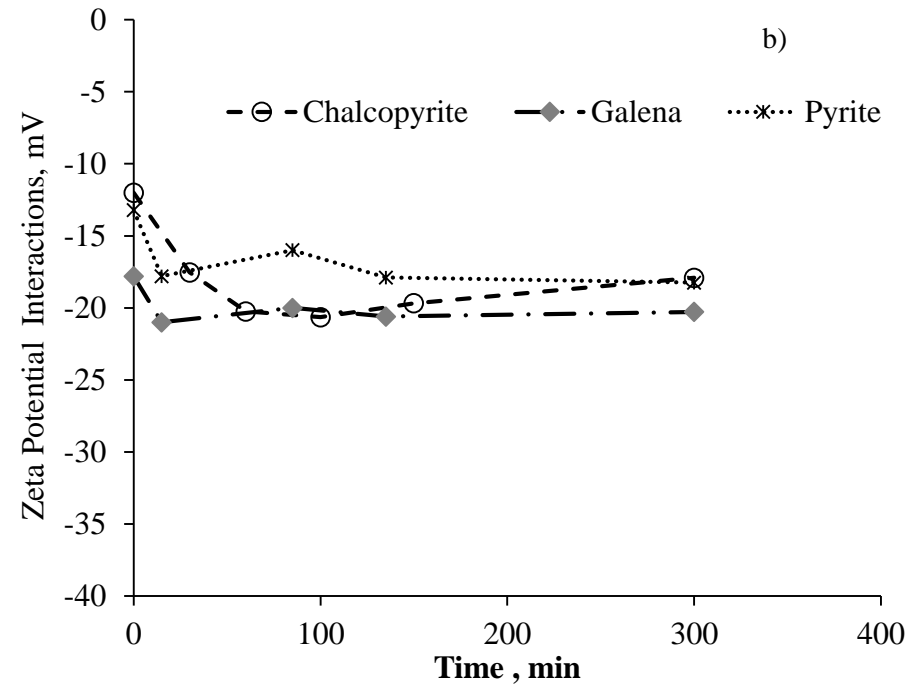
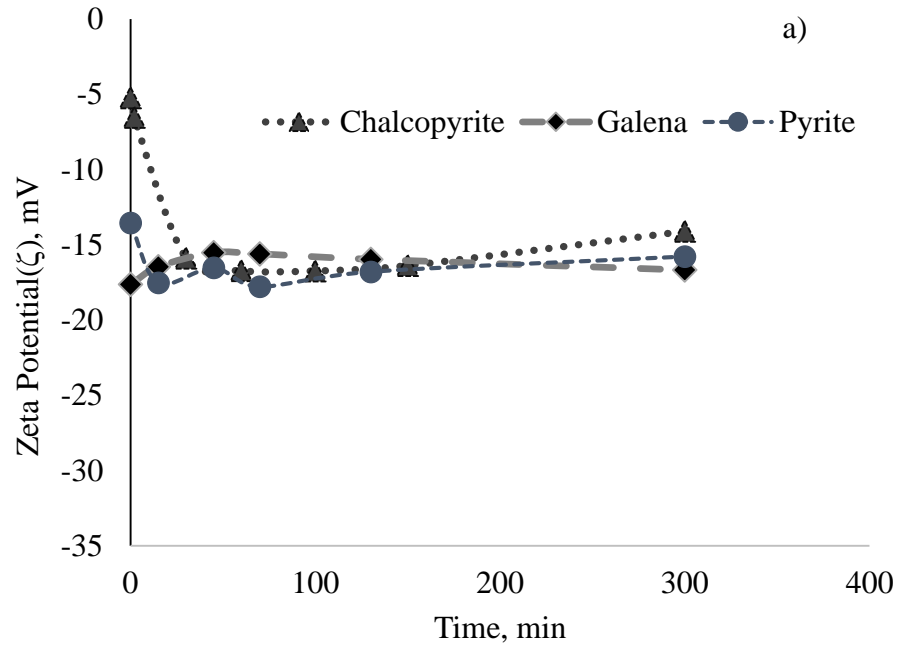


Figure 7. Zeta potential of *S. carnosus* bacteria with the mineral of interest. a) Before the interaction and b) after the interaction

Results

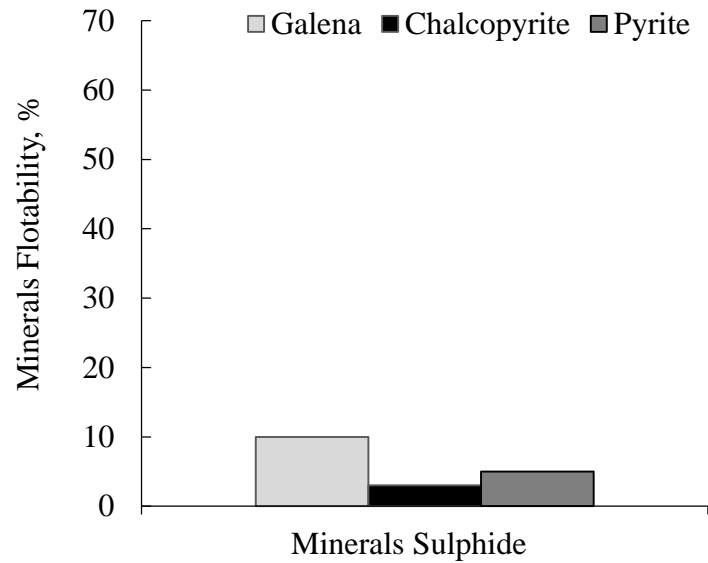


Figure 8. Floatability of the natural mineral at pH 9. Minerals conditioned in a 1×10^{-3} M solution of NaNO_3 at pH 9, in the absence of bacteria

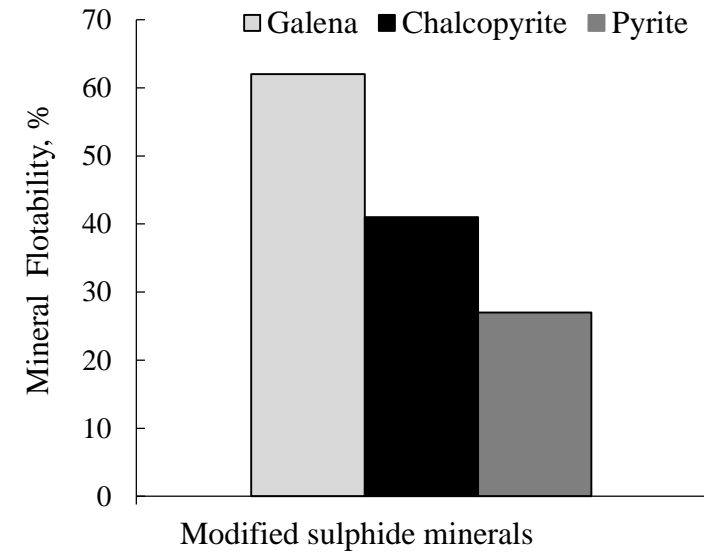


Figure 9. Floatability of bio-modified minerals. After interaction with *S.carnosus* bacteria at a pH of 9.

Annexes



ADHESIÓN DE BACTERIA

$$B_{Ad} = \frac{(B_0 - B)V}{wA_m}$$

Where: B_{Ad} is adhering bacteria (cell/m²); B_0 and B are the concentration of free bacteria at zero time and t , respectively (cell/ml); V is the volume of sample in ml; w is the exact weight of mineral in grams, A_m is the surface area determined by Coulter counter mineral

Conclusions



- From the results, it can be concluded that *S. carnosus* has a hydrophobic effect on minerals, resulting on its adherence on surface minerals and acting as a natural collector. The bio-modification generates different degrees of minerals hydrophobicity, which indicates the possibility to control the bio-collector for selective separation of minerals. The magnitude of the floatability of minerals showed the descending order as follows:
 - galena >> chalcopyrite > pyrite
- The sulfides mineral bio-modification mechanism was evaluated in the presence of *S. carnosus* and the results shows a physical mechanism, due to the electrostatic attraction generated between the bacterial surface charge or positive zeta mineral powder and negative surface charge (negative zeta potential).
- It is concluded that the usage of this strain are promising for environmentally friendly bioreagent, thus being applicable for the eco-friendly development of the bioflotation of sulfide minerals.

References



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Botero, A.E.C., Torem, M.L., de Mesquita, L.M.S., (2008). Surface chemistry fundamentals of biosorption of *Rhodococcus opacus* and its effect in calcite and magnesite flotation. *Minerals Engineering* 21, 83-92. <https://doi.org/10.1016/j.mineng.2007.08.019>

de Mesquita, L.M.S., Lins, F.F., Torem, M.L., (2003). Interaction of a hydrophobic bacterium strain in a hematite-quartz flotation system. *International Journal of Mineral Processing* 71, 3144. [https://doi.org/10.1016/S0301-7516\(03\)00028-0](https://doi.org/10.1016/S0301-7516(03)00028-0)

Dwyer, R., Bruckard, W.J., Rea, S., Holmes, R.J., (2012). Bioflotation and bioflocculation review: microorganisms relevant for mineral beneficiation. *Mineral Processing and Extractive Metallurgy (IMM Transactions section C)* 121, 65-71. <https://doi.org/10.1179/1743285512Y.0000000005>

Gericke, Y., Govender, M., 2011. Extracellular polymeric substances (EPS) from bioleaching systems and its application in bioflotation. *Minerals Engineering* 24 (11), 1122–1127. <https://doi.org/10.1016/j.mineng.2011.02.016>

Lopez, Leslie Y., Merma, Antonio G., Torem, Mauricio L., Pino, Gabriela H., (2015). Fundamental aspects of hematite flotation using the bacterial strain *Rhodococcus ruber* as bioreagent. *Minerals Engineering* 75, 63–69. <https://doi.org/10.1016/j.mineng.2014.12.022>

Pecina, E.T., Rodríguez, M., Castillo, P., Díaz, V., and Orrantia, E. (2009). Effect of *Leptospirillum ferrooxidans* on the flotation kinetics of sulphide ores. *Minerals Engineering* 22: 462-468. <https://doi.org/10.1016/j.mineng.2008.12.008>

Ramos Escobedo, G., Gallegos-Acevedo, P.M., López Saucedo, F.J., Orrantia-Borunda, E. (2012). Bioflotation of sulfide minerals with *Acidithiobacillus ferrooxidans* in relation to Cu-activation and surface oxidation. *Canadian Journal of Microbiology* 58, 1073-1083. <http://doi.org/10.1139/w2012-072>

Ramos-Escobedo, G.T., Pecina-Treviño, E.T., Bueno-Tokunaga, A., Concha-Guerrero, S.I., Ramos-Lico, D., Guerra-Balderrama, R., Orrantia-Borunda, E. (2016). Bio-collector alternative for the recovery of organic matter in flotation processes. *Fuel*, 176, 165-172. <https://doi.org/10.1016/j.fuel.2016.02.018>



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